LOYOLA COLLEGE (AUTONOMOUS), CHENNAI – 600 034

B.Sc. DEGREE EXAMINATION – CHEMISTRY

FOURTH SEMESTER - APRIL 2022

UCH 4602 – CHEMISTRY OF MATERIALS

Date: 23-06-2022 Dept. No. Time: 09:00 AM - 12:00 NOON

PART – A

Answer ALL questions.

- State the principle of sol-gel method. 1.
- 2. Distinguish between single and multi-walled carbon nanotubes.
- 3. State Curie Weiss law for ferrimagnets.
- What are cooper pair of electrons? 4.
- Define the term: Transducer. 5.
- Mention any two VOCs and their health effects. 6.
- 7. Define oligomers.
- 8. What are elastomers? Give an example.
- 9. Differentiate thermoplastics and thermosetting plastics with examples.
- How would you prepare Nylon-66? 10.

PART – B

Answer any EIGHT questions.

- Explain the synthesis of nanomaterials by inert gas condensation method. 11.
- 12. Discuss the properties of carbon nanotubes in brief.
- 13. What are metal excess defects? Explain.
- 14. Discuss the role of *p*-*n* junction in rectifiers.
- Explain the differences between ferro and paramagnetic materials. 15.
- 16. Distinguish between hard and soft magnets.
- What are biosensors? Explain the application of biosensors. 17.
- Explain the mechanism of VOC sensing of semiconductors. 18.
- 19. Describe the calendaring polymer processing technique.
- 20. Derive rate expression for free radical polymerization.
- How are Novolac and Bakelite prepared? Mention its applications. 21.
- Describe the vulcanization of rubber. 22.

PART – C

Answer any FOUR questions.

- 23. Draw the block diagram and explain the instrumentation of SEM.
- 24. Why do ferromagnets show spontaneous magnetisation? Explain.
- 25. Explain the mechanism of alcohol sensing by *n*-type and *p*-type semiconductors.
- 26. Discuss the generation of hydrogen using Honda Cell.
- 27a. Briefly explain the addition and condensation polymerization techniques with suitable examples.
- b. How will you synthesis Buna-S and Buna-N? Mention their applications.
- 28. Explain the synthesis of polyethylene using Ziegler-Natta catalyst.

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$(4 \times 10 = 40 \text{ Marks})$

 $(8 \times 5 = 40 \text{ Marks})$

Max.: 100 Marks

(10 x 2 = 20 Marks)